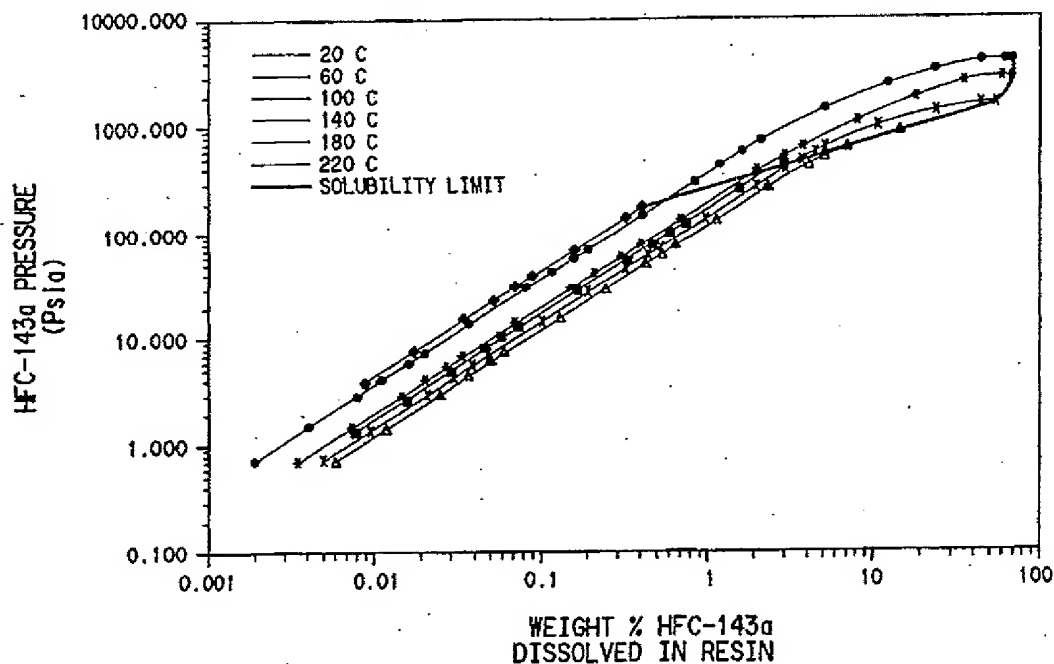




INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification ⁶ : C08J 9/14		A1	(11) International Publication Number: WO 98/03581
			(43) International Publication Date: 29 January 1998 (29.01.98)
(21) International Application Number: PCT/US97/13074		(81) Designated States: AL, AM, AU, AZ, BA, BB, BG, BR, BY, CA, CN, CU, CZ, EE, GE, HU, IL, IS, JP, KG, KP, KR, KZ, LC, LK, LR, LT, LV, MD, MG, MK, MN, MX, NO, NZ, PL, RO, RU, SG, SI, SK, SL, TJ, TM, TR, TT, UA, US, UZ, VN, YU, ARIPO patent (GH, KE, LS, MW, SD, SZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, TG).	
(22) International Filing Date: 23 July 1997 (23.07.97)			
(30) Priority Data: 60/022,575 24 July 1996 (24.07.96) US			
(71) Applicant (for all designated States except US): E.I. DU PONT DE NEMOURS AND COMPANY [US/US]; 1007 Market Street, Wilmington, DE 19898 (US).			
(72) Inventors; and		Published With international search report. Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.	
(75) Inventors/Applicants (for US only): CREAZZO, Joseph, Anthony [US/US]; 5 Quail Court, Wilmington, DE 19810 (US). HAMMEL, Howard, S. [US/US]; 5 Anna Avenue, Bear, DE 19701 (US). YORK, Robert, O. [US/US]; 20 Drummond Drive, Wilmington, DE 19808 (US).			
(74) Agent: SHIPLEY, James, E.; E.I. du Pont de Nemours and Company, Legal/Patent Records Center, 1007 Market Street, Wilmington, DE 19898 (US).			

(54) Title: CLOSED CELL THERMOPLASTIC FOAMS CONTAINING HFC-134



(57) Abstract

A closed cell styrenic or ethylenic thermoplastic resin foam is disclosed that was produced by means of a nonflammable blowing agent substantially free of halogen substituents other than fluorine and comprising HFC-134 (1,1,2,2-tetrafluoroethane).

FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AL	Albania	ES	Spain	LS	Lesotho	SI	Slovenia
AM	Armenia	FI	Finland	LT	Lithuania	SK	Slovakia
AT	Austria	FR	France	LU	Luxembourg	SN	Senegal
AU	Australia	GA	Gabon	LV	Latvia	SZ	Swaziland
AZ	Azerbaijan	GB	United Kingdom	MC	Monaco	TD	Chad
BA	Bosnia and Herzegovina	GE	Georgia	MD	Republic of Moldova	TG	Togo
BB	Barbados	GH	Ghana	MG	Madagascar	TJ	Tajikistan
BE	Belgium	GN	Guinea	MK	The former Yugoslav Republic of Macedonia	TM	Turkmenistan
BF	Burkina Faso	GR	Greece	ML	Mali	TR	Turkey
BG	Bulgaria	HU	Hungary	MN	Mongolia	TT	Trinidad and Tobago
BJ	Benin	IE	Ireland	MR	Mauritania	UA	Ukraine
BR	Brazil	IL	Israel	MW	Malawi	UG	Uganda
BY	Belarus	IS	Iceland	MX	Mexico	US	United States of America
CA	Canada	IT	Italy	NE	Niger	UZ	Uzbekistan
CF	Central African Republic	JP	Japan	NL	Netherlands	VN	Viet Nam
CG	Congo	KE	Kenya	NO	Norway	YU	Yugoslavia
CH	Switzerland	KG	Kyrgyzstan	NZ	New Zealand	ZW	Zimbabwe
CI	Côte d'Ivoire	KP	Democratic People's Republic of Korea	PL	Poland		
CM	Cameroon	KR	Republic of Korea	PT	Portugal		
CN	China	KZ	Kazakhstan	RO	Romania		
CU	Cuba	LC	Saint Lucia	RU	Russian Federation		
CZ	Czech Republic	LI	Liechtenstein	SD	Sudan		
DE	Germany	LK	Sri Lanka	SE	Sweden		
DK	Denmark	LR	Liberia	SG	Singapore		
EE	Estonia						

TITLE

CLOSED CELL THERMOPLASTIC FOAMS CONTAINING HFC-134

CROSS REFERENCE TO RELATED APPLICATIONS

5 This application claims the priority benefit of U.S. Provisional Application 60/022575, filed July 24, 1996.

FIELD OF THE INVENTION

10 The instant invention relates to using HFC-134 (1,1,2,2-tetrafluoroethane) in the production of thermoplastic foams.

BACKGROUND OF THE INVENTION

In general, thermoplastic foams are manufactured by mixing a volatile blowing agent with a molten thermoplastic resin under controlled conditions of temperature and pressure sufficient to form a plasticized, resin-blowing agent mixture and maintain the mixture in an unfoamed state. Thereafter, the mixture is extruded through a suitable die into a zone of lower pressure at a controlled temperature to obtain a substantially closed cell structure having a desired shape and form.

For satisfactory production of closed cell foams, the blowing agent's solubility should be sufficiently high, while the mixture is passing through the die into the lower pressure expansion zone, so that expansion proceeds smoothly to the closed cell state. Otherwise, if the solubility is too low the blowing agent vaporizes away prematurely — and so rapidly and to such an extent before the cell walls fully form — that the resulting foam contains a high proportion of ruptured cells (voids), which adversely affects the foam's utility, e.g., as insulating or load-bearing structural material. The blowing agents solubility in the resin is particularly critical when extruding through large cross-sectional dies. This is because the throughput of the extruding mixture is generally fixed in conventionally employed extruders and the back-pressure at the lip (opening) is low. The larger the area of the die opening the lower is the back-pressure exerted on the mixture and the greater is the number of voids in the resulting foam.

There are many conventional methods for making thermoplastic foams. Siraux et al, European Patent Application No. 0 406 206 A2 discloses polystyrenic foams and a method of making them involving a mixture of blowing agents comprising dichloromethane and one or more of a hydrochlorofluorocarbon (HCFC), a hydrofluorocarbon (HFC) and a fluorocarbon (FC), with dichloromethane comprising 5 to 25 weight percent of the mixture. The disclosed system has the disadvantage of utilizing and requiring environmentally objectionable (e.g., ozone

depleting) chloro-containing volatile substances as part of the blowing agent composition.

Suh et al, Canadian Patent No. 1,086,450, relates to polystyrenic closed cell insulating foams and their preparation, utilize a mixture of high permeability and low permeability blowing agents for foam production. These foams are undesirable because they require high proportions of objectionable flammable and/or chloro-containing high permeability blowing agents.

Suh, U.S. Patent No. 5,011,866, discloses blowing agents comprising at least 70 weight percent HFC-143a (1,1,1-trifluoroethane) or HFC-134a (1,1,1,2-tetrafluoroethane) for the preparation of polystyrenic insulating foams having small closed cells, low densities (1 to 6 pounds per cubic foot) and high dimensional stabilities among other properties. HFC-143a is objectionable for its flammability, and HFC-134a presents processing difficulties as indicated in York U.S. Patent Nos. 5,146,896 and 5,204,169.

Omire et al, U.S. Patent No. 5,145,606, disclose eleven categories of mixed blowing agents for foaming thermoplastics such as polystyrene and polyethylene. Included among the blowing agent mixtures are four that comprise a tetrafluoroethane, which may be HFC-134a and/or HFC-134 (1,1,2,2-tetrafluoroethane), mixed with one or more selected hydrochlorofluorocarbons (HCFC). Omire, however, teaches that HFC-134 and HFC-134a cannot be advantageously used alone as a blowing agent for thermoplastics, and tetrafluoroethane should be used in combination with at least one chloro-containing blowing agent.

Rubin et al, U.S. Patent No. 5,314,926, describe blowing agents for polystyrene and other non-isocyanurate based foamable plastics comprising 1,1,1,2,3,3,3-heptafluoropropane (HFC-227ea) in combination with one or more hydrocarbons or partially halogenated alkanes, which may contain chlorine substituents or be chlorine free. The disclosed system has the disadvantage of requiring HFC-227ea, a high HGWP (0.6) material, i.e. environmentally objectionable, as part of the composition.

Volcker et al, U.S. Patent No. 5,334,337, disclose high compression strength foam board prepared from polystyrene containing 5-16 weight percent of a blowing agent mixture containing an alcohol or ketone, carbon dioxide (CO₂) a C₃-C₅ hydrocarbon and fluorinated hydrocarbons. The disclosed technology is disadvantageous in that it includes at least one flammable component which is also a volatile organic compound (VOC); the usage of which is regulated in many countries.

Bartlett et al, U.S. Patent No. 5,182,040, disclose azeotropic and azeotropic-like binary compositions of HFC-134 with HFC-152a (1,1-difluoroethane), dimethyl

ether (DME) and selected halocarbons and hydrocarbons are useful as refrigerants, aerosol propellants and blowing agents for polymer foams.

SUMMARY OF THE INVENTION

5 We discovered that 1,1,2,2-tetrafluoroethane, HFC-134, is substantially as effective as HFC-152a as a blowing agent for thermoplastic resins with the added advantages over HFC-152a of affording nonflammability and lower permeability through thermoplastic resins. This is a surprising and unexpected result since those skilled in the art believe HFC-134 alone and an azeotrope thereof with HFC-152a
10 have essentially no effect on the solid polystyrene. Further, HFC-134 is superior to its isomer, HFC-134a, as a blowing agent for such foams. HFC-134 can exert a lower solution pressures in the extrusion step thereby permitting HFC-134 to be utilized in conventional foam manufacturing equipment.

The aforementioned discovery solves problems in this field by providing
15 improved thermoplastic insulating foams produced with the aid of effective environmentally friendly and nonflammable blowing agents, preferably in conventional equipment. The instant invention solves a further problem by providing foams suitable for use as insulating and structural elements wherein the blowing agents also exhibit acceptably low permeabilities through the thermoplastic resins.
20 sufficient for improved long-term insulating values. The instant invention solves these problems by providing an effective, environmentally friendly and nonflammable blowing agents for producing foam materials for food packaging and food service: normally by employing in conventional foam manufacturing equipment.

In one aspect, the invention comprises a closed, i.e., noninterconnecting, cell
25 thermoplastic resinous foam body produced by means of an environmentally friendly nonflammable and low permeability polyfluorocarbon blowing agent composition comprising more than about 70 weight percent HFC-134 and being substantially completely devoid of components having halogen substituents other than fluorine. The instant invention can be employed to obtain closed-cell foams having a wide
30 range of characteristics. The cells of the foam typically have an average cell size of from about 0.1 to about 1.5 millimeters (mm) when measured across a minimum cross-sectional dimension of the body. The foam body typically has a minimum cross-section dimension (thickness) of at least about 0.04 inch (1 mm), a cross-sectional area of at least about 2 square inches (13 square centimeters) and a density
35 of from about 0.75 to 15 pounds per cubic foot (pcf or 12 to 240 kilograms per cubic meters). Normally, the foam has a thickness of at least about 0.5 inch (1.27 cm), typically at least about 1.0 inch (2.54 cm), and usually at least about 1.5 inches (3.9 cm). The cross-sectional area of the foam body can be at least about 4 square inches

(26 square centimeters), usually at least about 8 square inches (52 square centimeters) and normally at least about 16 square inches (104 square centimeters). While the density of the foam body is dependent upon many variables, the density is typically at least about 1.5 pcf (24 Kg/m³), normally at least about 3 pcf (48 Kg/m³) and usually at least about 6 pcf (96 Kg/m³).

The foam body can be comprised of any suitable thermoplastic resin. While HFC-134 can be employed with a virtually unlimited array of resins, the thermoplastic resin is normally, polystyrenic; but, polyethylenic or polypropylenic can also be employed. Further details regarding the composition of the foam body as well as the components/systems that are used for forming the foam body can be found in "Modern Plastics Encyclopedia '92," Volume 68, Number 11; Chapters on: "Primary Processing," "Chemicals and Additives"; the disclosure of which is hereby incorporated by reference.

The foam body is normally produced by (a) forming a substantially homogeneous molten mixture of the normally solid thermoplastic resin and an effective foaming amount of the blowing agent at a nonfoaming elevated temperature and pressure, (b) extruding the mixture through a die into an expansion zone at a controlled foaming temperature and reduced pressure at a controlled rate effective to form a closed cell substantially void free foam body, (c) allowing the foam body to cool and increase in viscosity at temperatures and pressures such that there is obtained a substantially closed cell foam body having cell size dimensions and density as defined above at ambient temperatures and atmospheric pressure. The specific process conditions for obtaining the substantially closed cell body are understood by a skilled person in this art.

In one aspect the invention comprises sheets of relatively low density closed cell foam body suitable for use as a thermoformable starting material for producing food-packaging materials. One suitable technique for making such foams are described in U.S. Patent No. 5,204,169; the disclosure of which is hereby incorporated by reference. Another aspect comprises a relatively thick and high density foam body suitable as insulation, for example, in the form of boardstock in building constructions. In a further aspect the invention comprises higher density closed cell insulating foams suitable for use as structural members. The insulating foam bodies of the invention are further characterized as exhibiting low blowing agent permeabilities, e.g., low loss of blowing agent, thereby improving thermal insulation value of the foam over time as measured by its K-factor.

The blowing agent composition for use in this invention comprises HFC-134 alone or as substantially nonflammable environmentally friendly mixtures thereof containing less than about 30 weight percent of other compatible blowing agents that

are devoid of halogen substituents other than fluorine. Such blowing agents normally have 1 to 2 carbon atoms and are not classified as photochemically reductive VOCs (volatile organic compounds). That is, the blowing agent composition is substantially nonflammable, has zero ozone depletion potential (ODP) and low halocarbon global warming potential (HGWP) or "greenhouse effect". In some cases, it may be desirable to employ HFC-134 in conjunction with one or more additive or co-blowing agents selected from the group consisting of HFC-134a (1,1,1,2-tetrafluoroethane, $\text{CF}_3\text{CCH}_2\text{F}$), HFC-152a (1,1-difluoroethane, CF_2HCH_3), HFC-143a (1,1,1-trifluoroethane, CF_3CH_3), HFC-32 (difluoromethane, CF_2H_2), HFC-125 (pentafluoroethane, $\text{CF}_3\text{CF}_2\text{H}$), among others. Normally, in terms of weight percent the amount of additive blowing agent corresponds to less than about 22, usually less than about 13, and typically substantially nil, based on the total weight of the blowing agent composition. The inventive blowing agent includes azeotropic and azeotrope-like mixtures of HFC-134 with said co-blowing agents. An example of such azeotropes includes the flame resistant and nonflammable mixtures of HFC-134 with HFC-152a within the range of 78 weight % or more of HFC-134 and 22 weight % or less of HFC-152a, and normally within the range of 87 weight % or more of HFC-134 and 13 weight % or less of HFC-152a. Examples of such nonflammable mixtures are also disclosed in the aforementioned Watanabe et al, EP 483 573A1 and Bartlett et al. references.

By an "effective blowing agent foaming amount" it is meant an amount sufficient to result in foam body having the properties described herein. Normally, an effective amount will range from about 1 to about 30 weight percent based on the total weight of the resin blowing agent composition, usually about 2 to about 20, and typically about 2 to about 10 weight percent. The specific effective amount of the blowing agent will depend upon the particular type and grade of resin being employed and the desired resultant properties.

By a "closed cell substantially void-free foam body" it is meant a foam with a substantially uniform cell structure which has greater than about 50% closed cells, preferably greater 90% closed cells as measured by ASTM D2856-70.

By "nonflammable" it is meant to refer to satisfying the criteria set forth in ASTM E6871-85 with modifications listed in draft form, Nov. 1993, by ASTM Committee E27 or by ASTM E918; hereby incorporated by reference.

By "low halocarbon global warming potential (HGWP) effect," it is meant a blowing agent having an HGWP value of about 0.50 or less, usually less than about 0.32 and typically less than about 0.30 as determined by the method described by D. A. Fisher et al, NATURE 1990, 344, p. 513; hereby incorporated by reference.

Cell sizes of the foam are determined by ASTM method D2842-69, and foam dimensional stabilities by ASTM D2126/C758.

Foam densities are determined by ASTM D1622-83.

The invention is based on the discovery that contrary to expectations HFC-134: (1) is superior to its isomer HFC-134a as blowing agent for thermoplastic foams. in that it and mixtures thereof exert substantially lower solution pressures (higher solubilities in the resin phase) over the range of temperatures and pressures employed in extruding the compositions as foams such that HFC-134 may be used in conventional equipment without modification to produce substantially void-free closed cells; (2) alone or as mixtures thereof produce high quality closed cell foams having desirably small cell sizes over a wide range of densities, dimensional stability and low permeability through thermoplastic film, thus providing long-lived insulating and structural foams; (3) is superior to HFC-143a and HFC-152a, which are recommended in the art for use in producing thermoplastic insulating foams, in that HFC-134 and mixtures thereof have lower permeability through thermoplastic film thereby providing improved insulating characteristics while offering the added advantages of being nonflammable; (4) if alone or in combination with other environmentally friendly hydrofluorocarbon blowing agents can be an acceptable replacement of such heretofore commercially employed blowing agents for thermoplastic foams such as CFC-11, CFC-12, HCFC-22 and HCFC-142b.

BRIEF DESCRIPTION OF THE DRAWINGS

Figure 1 graphically represents the solubilities in a typical normally solid polystyrenic resin of the inventive HFC-134 in comparison to HFC-134a, HFC-152a and of HFC-143a over a wide of range of temperatures embracing extrusion temperatures useful in the foam-producing process of the invention.

Figure 2 graphically represents the interaction parameters of HFC-134, HFC-134a, HFC-152a and HFC-143a with polystyrene over a temperature range of 20 to 220°C.

Figures 3, 4, 5, and 6 are log-log plots that relate to the solubility of HFC-134, HFC-134a, HFC-152a and HFC-143a, respectively, as superheated vapor in polystyrene resin over a 20 to 220°C temperature range.

DETAILED DESCRIPTION

One aspect of the invention comprises a substantially solid closed cell thermoplastic insulating foam body wherein the cells are substantially completely filled with a substantially nonflammable blowing agent comprising more than about 70 weight percent of HFC-134, the balance, if any, of the blowing agent comprising

one or more components that are devoid of halogen substituents other than fluorine, and have low HGWP values. Examples of suitable blowing agent components for use with HFC-134 include hydrofluorocarbons comprising at least one of HFC-134a, HFC-143a, HFC-152a, HFC-32, HFC-125, azeotropic and azeotrope-like mixtures with HFC-134, e.g., as described above, among others. Other acceptable blowing agent components that may be used in conjunction with HFC-134, with or without the other HFC components, comprise at least one of N₂, CO₂, argon, other rare gases, among others. Hydrocarbons and other VOC compounds are desirably absent.

Normally, the HFC-134 content of the blowing agent composition is at least about 78% by weight, usually at least about 87, and desirably about 100 percent. Typically, the average cell size is less than about 1.5mm, and normally not more than about 1.2 mm. When the foam is employed as an insulating body, the thickness can be between about 0.04 and about 6 inches (1 to 152 mm), and the foam density from about 0.75 to about 15.0 pounds per cubic foot (12 to 240 kilograms per cubic meter). Consequently, the foam of the instant invention provides small cells, is obtained by using a low permeability environmentally friendly blowing agent(s), and provides long-term insulating properties to thermoplastic foam bodies while excluding release of ozone-depleting halocarbons and smog-producing volatile organic compounds into the atmosphere.

The polystyrenic resinous component of the invention foam products, sometimes referred to in the art as "styrenic resin", may vary widely in chemical composition. Broadly included as resins that can be employed in the instant invention are solid thermoplastic polymers of one or more polymerizable alkenylaromatic compounds. The polymer or copolymer comprises in chemically combined form at least one alkenylaromatic compound having the general formula $\text{Ar-C(R)} = \text{CH}_2$, wherein Ar represents an aromatic hydrocarbon radical of the benzene series, typically phenyl, and wherein R normally represents hydrogen (preferably) or a methyl radical. While any suitable alkenyl aromatic resins can be employed, examples comprise the solid homopolymers of styrene, alpha-methylstyrene, o-methylstyrene, m-methylstyrene and p-methyl styrene; also the solid copolymers of two or more such alkenylaromatic compounds, among others. The polymers may also include in copolymerized form relatively minor proportions of other polymerizable olefinic compounds, such as, for example, methyl methacrylate, acrylonitrile, methacrylic acid, acrylic acid, maleic anhydride, among others. Without wishing to be bound by any theory or explanation, it is believed that the non-styrenic monomers can contribute via their oxygen and nitrogen containing moieties polymer sites for solvating HFC-134 (and other hydrofluorocarbon components of the blowing agent) during the step of mixing polystyrenic resins blowing agent blends prior to foaming.

A desirable polystyrenic resin comprises the solid homopolymer, polystyrene, due to its low cost and readily availability. Other thermoplastic resins can be foamed with the HFC-134 based blowing agent composition, including polyolefinic compounds, for example, polyethylene, polypropylene, mixtures thereof, among others.

5 The thermoplastic foam bodies of this invention are conveniently produced by using conventional equipment comprising an extruder and associated means for (1) melting the resin; (2) homogeneously blending the blowing agent composition with the melt to form a plasticized mass at nonfoaming temperatures and pressures; (3) passing the plasticized mass at a controlled rate, temperature and pressure through a
10 die having a desired shape, e.g., slit die for producing rectangular slabs of foam board having desired thickness and surface area, into an expansion zone; (4) allowing the extrudate to foam in the expansion zone maintainable at suitable temperatures and low pressures; (5) maintaining the expanding extrudate under such temperatures and pressures for a time sufficient for the viscosity of the extrudate to increase such that
15 the cell size and density of the foam remain substantially unchanged and substantially free of ruptured cells at ambient temperature, e.g., 25°C and atmospheric pressure; and (6) recovering the extruded foam body.

In one aspect of the invention, the foaming amount of the blowing agent is in the range of from about 1 to about 30 weight percent based on the total weight of the
20 resin plus blowing agent mixture, typically about 2 to about 20 weight percent, and normally about 2 to about 10 weight percent. The relatively lower the concentration of blowing agent, the greater the density of the resulting foam. The proper amount of blowing agent or resultant characteristics of the foam for any desired end-use are readily determined by a skilled person in this art who has reviewed and understood the
25 instant invention. The resin is melted at a temperature of about 200 to about 235°C depending upon the grade employed, and at nonfoaming pressures of about 600 psig or higher. The plasticized resin-blowing agent mixture is cooled under nonfoaming pressure to a temperature of about 115 to 150°C, normally 130°C, and extruded into the expansion zone at or below ambient temperature and at or below atmospheric
30 pressure.

When preparing foams of this invention, it is often desirable to add a nucleating agent or other additive into the resin. Nucleating agents serve primarily to increase cell count and decrease cell size in the foam, and may be used in an amount of about 0.1 to about 4 parts by weight per 100 parts by weight of the resin. Typical
35 nucleating agents comprise at least one member selected from the group consisting of talc, sodium bicarbonate-citric acid mixtures, calcium silicate, carbon dioxide, among others. Other additives often also incorporated into the resin include, for example,

colorants, antioxidants, lubricants, stabilizers, fire retardants, among others, depending upon the end use of the resin.

Representative foamed products that can be made in accordance with the present invention comprise: (1) polystyrene foam sheet for the production of disposable thermoformed packaging materials, e.g., as disclosed in York U.S.P.N. 5,204,169; (2) extruded polystyrene foam boards for use as residential and industrial sheathing and roofing materials, which may be from about 0.5 to 6 inches (1.25 to 15 cm) thick, up to 4 feet (122 cm) wide, with cross-sectional areas of from 0.17 to 3 square feet (0.016 to 0.28 square meter), and up to 27 feet (813 meters) long, with densities of from about 1.5 to 10 pounds per cubic foot (pcf) (25 to 160 kilograms per cubic meter (kg/m^3)); (3) expandable foams in the form of large billets which may be up to about 2 feet (61 cm) thick, often at least 1.5 feet (46 cm) thick, up to 4 feet (1.22 meters) wide, up to 16 feet (4.8 meters) long, having a cross-sectional area of about 2 to 8 square feet (0.19 to 0.74 square meter) and a density of from 6 to 15 pcf (96 to 240 kg/m^3). Such foamed products are more fully described by Stochdopole and Welsh in the Encyclopedia of Polymer Science and Engineering, vol. 16, pages 193-205. John Wiley & Sons, 1989; hereby incorporated by reference.

Certain aspects of the invention are illustrated by the Figures. The accompanying figures illustrate the suitability of HFC-134 as blowing agent for producing polystyrenic foam bodies, the data presented therein were obtained with the aid of the well-known Flory-Huggins equation (Flory, P.J.: "Principles of Polymer Chemistry"; Cornell University Press, Ithaca, N.Y.), discussed below.

Referring now to the Figures, Figure 1 is a plot of weight% material soluble in polystyrene resin vs. temperature which shows that the solubility of HFC-134, in comparison to other HFCs, in a representative polystyrene resin (having a glass transition temperature of 85°C and a melting point of 105°C) increases with increasing temperatures, as measured at the mixtures' equilibrium pressures. These equilibrium pressures relate to the point in the extrusion process where blowing agent is mixed with resin; foaming cannot occur at these pressures. Figure 1 shows that the solubility of HFC-134 and HFC-152a in the selected resin are similar over the entire temperature range, and that their solubilities are significantly greater than those of HFC-134a and HFC-143a. HFC-134 and HFC-152a also show superior solubilities under extrusion conditions; data demonstrating such solubility are illustrated in Figures 3, 4, 5 & 6 and are abstracted below in Tables 1 and 2.

Referring now to Figures 3-6, Figures 3-6 relate to the solubility of superheated HFC-134 vapor in polystyrene resin at the following selected temperatures 20°, 60°, 140°, 180° and 220°C at pressures as high as about 3000 psia. The data presented in Figures 3-6 were determined through use of the Flory-Huggins

equation for phase equilibria in polymer solutions. This equation uses values for the following parameters [(1) - (4)], which were obtained as follows:

- (1) liquid densities of the blowing agents, g/cc, were derived by using standard methods, drawing a straight line through the liquid density data (based on National Institute of Standards and Technology, NIST, experimental data; with equations based on the Modified Benedict-Webb-Rubin, MBWR, equation of state) from -50°C to 50°C and calculating the liquid density as a function of temperature. The density (d) equation recites: $d=AT+B$, wherein A is the slope of the line, B is the constant of the line, and T temperature (°C). The liquid densities are (grams/cubic centimeter, g/cm³):

		<u>Liquid Densities</u>				
		<u>50°C</u>	<u>85°C</u>	<u>100°C</u>	<u>120°C</u>	<u>150°C</u>
15	HFC-134	1.214	1.104		0.994	0.888*
	HFC-134a	1.108	0.983			0.749
	HFC-152a	0.819		0.655	0.572	
	HFC-143a	0.804		0.645	0.553**	0.460
	Resin***	1.056	1.051	1.041	1.025	1.011
20	* 160°C					
	** 125°C					
	*** Arco Dylene® 8G Polystyrene, t.g.=85°C, m.p. 105°C					

- (2) The densities of the superheated vapors of the blowing agents, which are tabulated below, were calculated from the known vapor data of their thermodynamic properties (based on National Institute of Standards and Technology, NIST, experimental data; with equations based on the Modified Benedict-Webb-Rubin, MBWR, equation of state).

		<u>Superheated Vapor Densities (g/cm³)</u>				
		<u>50°C</u>	<u>85°C</u>	<u>100°C</u>	<u>120°C</u>	<u>150°C</u>
30	HFC-134	0.04379	0.04497		0.03830	0.04867*
	HFC-134a	0.03390	0.02975			0.03158
	HFC-152a	0.02495		0.02281	0.03850**	0.02942
	HFC-143a	0.04611	0.04386		0.03204**	0.04349
35	* 160°C					
	** 125°C					

(3) The activity coefficients for the blowing agents that are tabulated below:

		<u>Activity Coefficients</u>			
	<u>Temp. °C</u>	<u>HFC-134</u>	<u>HFC-134a</u>	<u>HFC-152a</u>	<u>HFC-143a</u>
5	50	0.877	0.605	0.728	0.537
	85	0.447	0.273		
	100			0.277	0.238
	120	0.218			
	125				
10	150		0.112	0.174	0.133
	160	0.116			

These coefficients were determined by dividing the experimental blowing agent partial pressure by the saturated vapor pressure at that temperature.

(4) The interaction parameters, X₁, were determined at four temperatures, and are presented graphically as a function of temperature in Figure 2 by means of the following equation and equation constants:

$$X_1 = A/T + B \text{ (where } T = ^\circ K = ^\circ C + 273.2 \text{)}$$

	<u>Compound</u>	<u>A</u>	<u>B</u>
25	HFC-134 (<85°C)	2553.95	-5.75256
	HFC-134 (>85°C)	2075.829	-4.44256
	HFC-143a (<85°C)	4641.477	-11.5113
	HFC-143a (>85°C)	2450.138	-5.40252
	HFC-152a (<85°C)	1944.713	-4.51260
	HFC-152a (>85°C)	2013.858	-4.64595

Representative X₁ values are:

	<u>Temp. °C</u>	<u>HFC-134</u>	<u>HFC-134a</u>	<u>HFC-152a</u>	<u>HFC-143a</u>
35	50	2.2	2.80	1.50	2.90
	100	1.1	1.50	0.75	1.20
	125	0.75	1.10	0.40	0.75
	150	0.50	0.80	0.10	0.40
	200	-0.05	0.20	0.40	-0.25

Referring now to Figure 2, these data permit calculating the limiting and superheated solubilities of the blowing agents via the Flory-Huggins equation, utilizing the values of (1)-(4) above. Such data are presented graphically in Figures 3.

4, 5 and 6. Figures 3-6 are plots of "weight% Blowing Agent Dissolved in Resin vs. Pressure of Solution" as a function of temperature. Representative solubilities of the blowing agents in the given resin at various extrusion process temperatures at atmospheric pressure (14.7 psia), are listed below in Table 1:

Table 1 WT% Blowing Agent Dissolves in Resin

Temp.°C	HFC-134	HFC-134a	HFC-143a	HFC-152a
200	0.15	0.08	0.05	0.15
180	0.2	0.1	0.07	0.2
140	0.3	0.15	0.1	0.3
100	0.4	0.19	0.13	0.4
60	0.5	0.2	0.08	0.5
20	0.6*	0.16*	0.04	0.8

*.25°C

The Table 1 data indicate that during the extrusion foaming stage of foam production the solubility, respectively, of HFC-134 and HFC-152a in the resin per se, as distinct from the amount of such material in the cells of the resultant foam, progressively increase with decreasing temperature at a given foaming pressure. The same trend is evident upon examining the data from the Figures at the other pressures, e.g., 100 psia.

This resin solubility effect indicates that a better, more uniform foam will result. The HFC-134 (and HFC-152a) solubilities in the resin increase, i.e., solution pressures decrease, with decreasing temperature during extrusion. These characteristics are advantageous by providing conditions for foam formation which are conducive to closed cell production over a wide range of densities while minimizing the possibility of blown/ruptured cells.

The data in these Figures, show that the HFC-134/HFC-134a solubility ratio is substantially steady at a ratio of about 2/1 over the 200 to 100°C temperature range during which the resin-blowing agent composition is substantially molten. The solubility ratio, however, increases at temperatures of about 60°C and 20°C, which are below the melting and glass transition temperatures of the resin. These solubilities are surprising and unexpected, and when translated into the closed cell foam product correspond to a wide range of densities with fewer blow/ruptured cells.

The following Table 2 presents similar solubility data that demonstrates the superiority of HFC-134 and HFC-152a over the other blowing agents at an extrusion temperature of 180°C, in terms of the (1) higher limiting, i.e., maximum solubilities at the extrusion temperature, (2) the lower vapor pressures

exerted by HFCs 134 and 152a at a 10 weight percent solubility in the resin, and; (3) the more favorable solubilities at 1 atmosphere pressure at both the extrusion temperature of 180°C and the final at rest temperature of 25°C.

5

Table 2

	<u>Blowing Agent</u>	<u>Extruder Temp °C</u>	<u>Limiting Solubility (wt % HFC)</u>	<u>at Extruder Temp for 10% Load (psia)</u>	<u>Solubility (wt % HFC) at 1 Atm of HFC Pres. Ext Temp 25°C</u>	
10	HFC-134	180	81	630	0.2	0.54
	HFC-134a	180	69	1000	0.11	0.16
	HFC-143a	180	69	1370	0.07	0.04
	HFC-152a	180	83	620	0.2	0.76

15

Similar results are obtained at higher and lower extrusion temperatures, e.g., 200° and 130°C.

Table 3

20 Solution Pressure of Blowing Agents in Polystyrene Resin at 140°C.

	<u>Blowing agent</u>	<u>Concentration (wt% in resin*)</u>	<u>Solution Pressure (psia)</u>
	HFC-134	10.00	450
25	HFC-134a	10.00	740
	HFC-143a	8.23	810
	HFC-152a	6.47	340

* wt% adjusted to achieve molar equivalency

30

Employing a procedure identical to that detailed above on Quantum "Petrothene" grade low density polyethylene, the following Table 4 presents similar solubility data that demonstrates the superiority of HFC-134 over other fluorinated blowing agents at an extrusion temperature of 180°C, in terms of the (1) higher limiting, i.e., maximum solubilities at the extrusion temperature, (2) the lower vapor pressure exerted at 10 weight percent solubility in the resin, and; (3) the more favorable solubility at 1 atmosphere pressure at both the extrusion temperature of 180°C and the final at rest temperature of 25°C.

35

Table 4

Blowing Agent	Extruder Temp °C	Limiting Solubility (wt % HFC)	at Extruder Temp for 10% Load (psia)	Solubility (wt % HFC) at 1 Atm of HFC Pres.	
				Ext Temp	25°C
HFC-134	180	81	621	0.20	0.64
HFC-161*	180	89	600	0.24	0.22
HCFC-114**	180	78	324	0.38	2.01
HFC-134a	180	74	828	0.14	0.22

* HFC-161 is 1-fluoroethane (CFH_2CH_3)

**HCFC-114 is 1,2-dichloro-1,1,2,2-tetrafluoroethane ($\text{CClF}_2\text{CClF}_2$)

Example 1

The following example serves to illustrate the ability to use HFC-134 to produce a polystyrene insulation foam with fine, uniform cell structure, long-term insulation value, and good dimensional stability.

The extruder employed was designed for use with CFC-12 (CCl_2F_2) and was converted to a blowing agent blend of HCFC-142b (CFCl_2CH_3)/HCFC-22 (CHClF_2) 60/40 wt.%. The HCFC-142b/HCFC-22 blend exhibits solution pressure in polystyrene similar to CFC-12.

The example 1 data reveals that HFC-134 performs very similar to HCFC-142b/HCFC-22 blend (note die pressure, foam thickness, foam width, and foam density). With HFC-134a, the die gap had to be closed to control the process (from 1.9 mm to 1.7 mm), which yielded a higher operating pressure (2483 psig). It was also necessary to reduce the resin melt temperature (from 129 °C to 116 °C) to reduce premature foaming in the die. However, even with these changes, HFC-134a blown foam was heavy (43 kg/cubic meter), had a rough surface, and did not achieve the required thickness and width.

Foam trials to produce polystyrene foam insulation were conducted using a commercial tandem extruder equipped with an experimental die, designed for higher pressure operation.

Primary extruder diameter = 120 mm.

Secondary extruder diameter = 200 mm.

Polystyrene resin = Shell NX606 general purpose, 2.5 melt index.

Nucleator = Magnesium silicate talc.

	<u>Trial 1</u>	<u>Trial 2</u>	<u>Trial 3</u>
<u>Blowing agent</u>	<u>22/142b</u>	<u>HFC-134a</u>	<u>HFC-134</u>
	40/60 wt %		
5			
Primary extruder (rpm)	70	70	70
Primary extruder (amps)	264	270	267
Extrusion rate (kg/hr)	430	422	438
Blowing agent rate (kg/hr)	46.5	45.8	45.1
10 Blowing agent concentration (wt.%)	10.82	10.86	10.31
Nucleator concentration (wt.%)	0.6	1.16	0.8
15 Secondary extruder speed (rpm)	4.9	4.9	4.9
Secondary extruder (amps)	102	122	112
Die pressure (psig)	1484	2483	1645
Melt temperature (deg. C.)	129	116	127
Die gap (mm)	1.9	1.7	1.9
20 Die width (mm)	100	100	100
Foam thickness (mm)	52	44	50
Foam width (mm)	317	249	285
Foam density (kg/cubic meter)	30.5	43	32.5
25 Comments	Excellent foam	Too thin Foaming in die Rough surface	Excellent foam

THE FOLLOWING IS CLAIMED:

1.) A process for producing a closed cell substantially void-free styrenic or ethylenic resin foam by means of a substantially nonflammable hydrofluorocarbon blowing agent composition comprising more than 70 weight percent 1,1,2,2-tetrafluoroethane (HFC-134) and being devoid of components having halogen substituents other than fluorine, said cells having an average cell size of from about 0.1 to about 1.5 mm and said foam having a density of from about 0.75 to 15 pounds per cubic foot (12 to 240 kilograms per cubic meter), comprising the steps of:

(a) forming a substantially homogenous molten mixture of a normally solid form of said thermoplastic resin and an effective foaming amount of the blowing agent at a nonfoaming elevated temperature and pressure,

(b) extruding the mixture through a die into a zone at a foaming elevated temperature and pressure and at a controlled rate effective to obtain a closed cell foam body, and

(c) allowing the foam body to cool and increase in viscosity at temperatures and pressures such that there is obtained a substantially rigid closed cell foam body having said cell size and density, wherein

said process is carried out in conventional apparatus for preparation of thermoplastic foam in which conventional thermoplastic foam blowing agents, such as dichlorodifluoromethane (CFC-12) or mixtures of 1-chloro-1,1-difluoroethane (HCFC-142b)/chlorodifluoromethane (HCFC-22), are employed.

2. The process of Claim 1 wherein said foam is produced with about 2 to about 10 weight percent of the blowing agent based on the total weight of the resin-blowing agent composition, and the foam density is at least 1.5 pcf (24 kg/m³).

3. The process of Claim 2 wherein the foam density is at least 6 pcf (96 kg/m³):

4. The process of Claim 1 wherein the blowing agent composition further comprises at least one member selected from the group consisting of HFC-134a, HFC-152a, HFC-143a, HFC-32 and HFC-125 in an amount totalling less than about 30 weight percent.

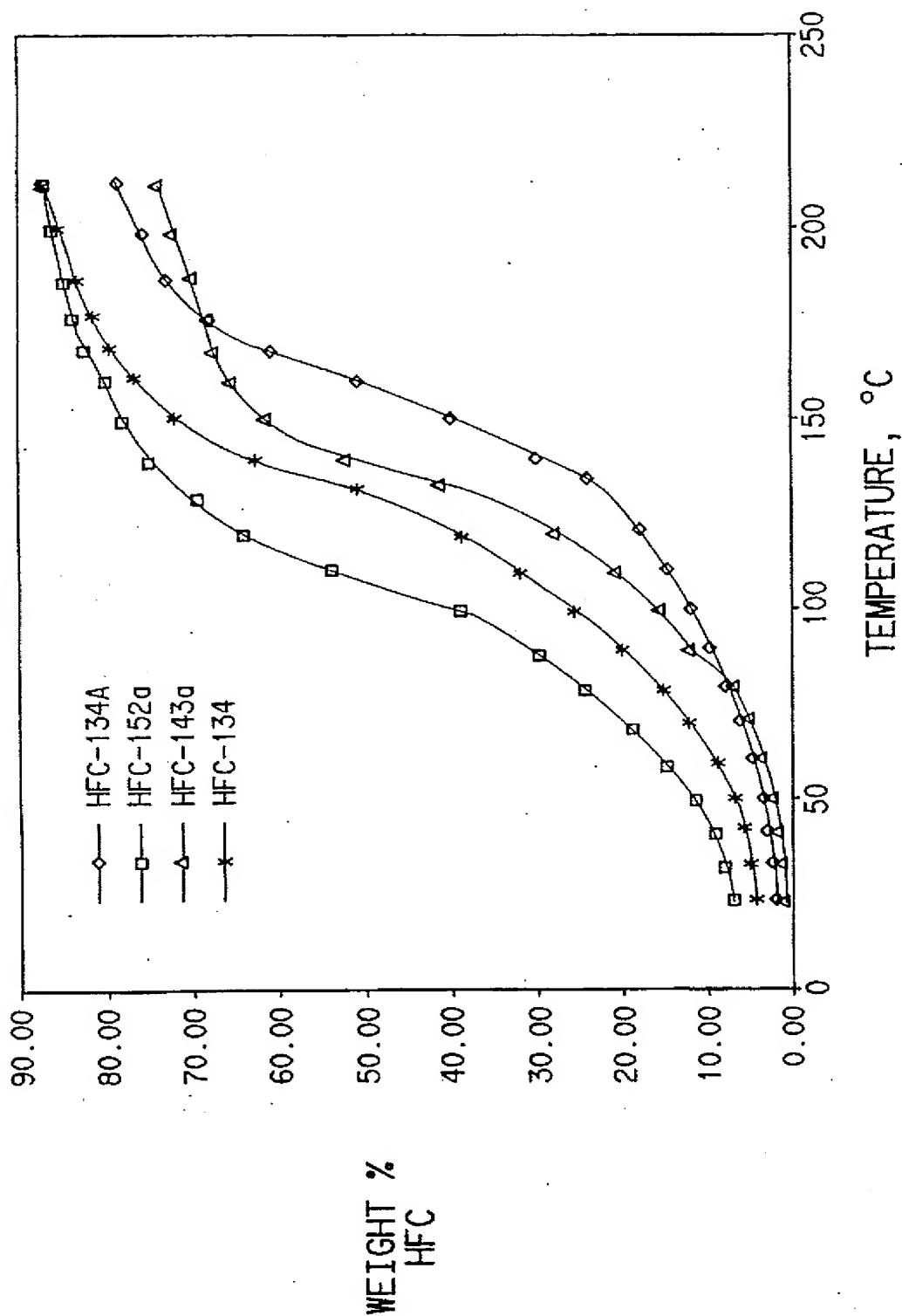
5. The process of Claim 1 wherein the blowing agent composition comprises at least 78 weight percent HFC-134.

6. The process of Claim 1 wherein the blowing agent comprises at least about 87 weight percent HFC-134.

7. The process of Claim 1 wherein the blowing agent consists essentially of HFC-134.

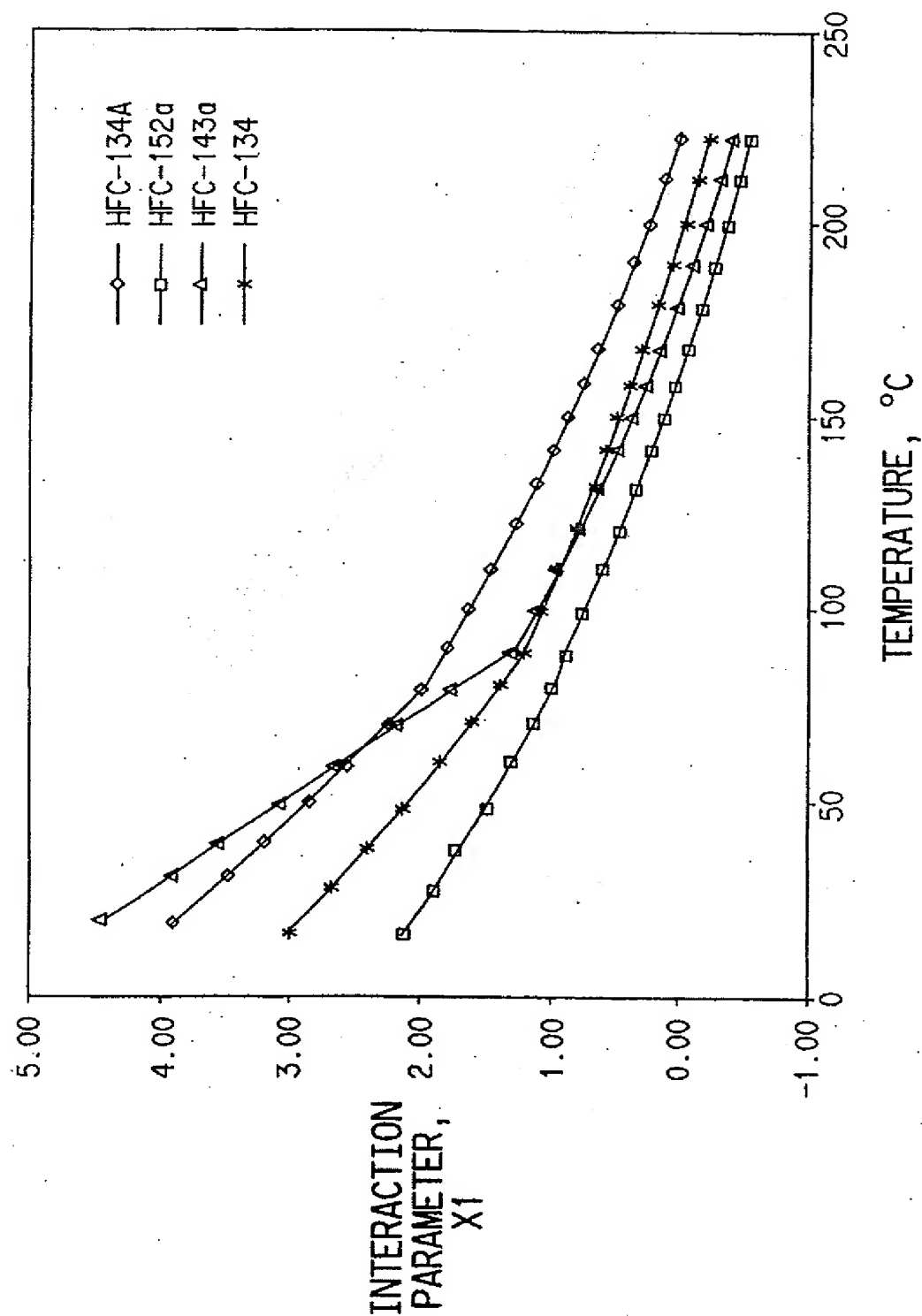
8. The process of Claim 1 wherein, when compared at an identical temperature and prior to extrusion through said die, the ratio of vapor pressure of said molten mixture to the vapor pressure of an identical molten mixture containing HFC-134a in place of HFC-134 is less than the ratio of normal vapor pressures of HFC-134 to HFC-134a.
- 5

FIG. 1



2/6

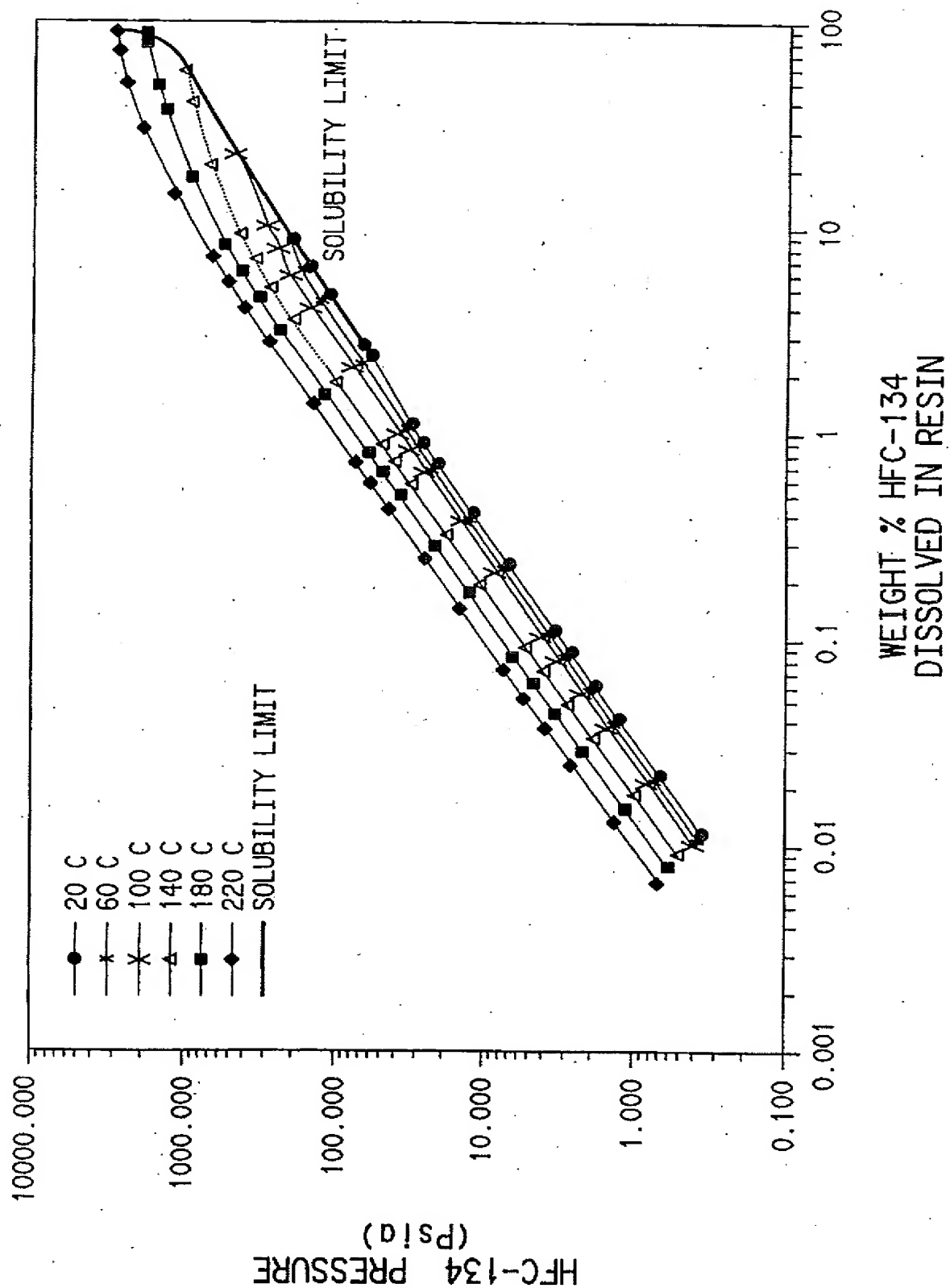
FIG. 2



SUBSTITUTE SHEET (RULE 26)

3/6

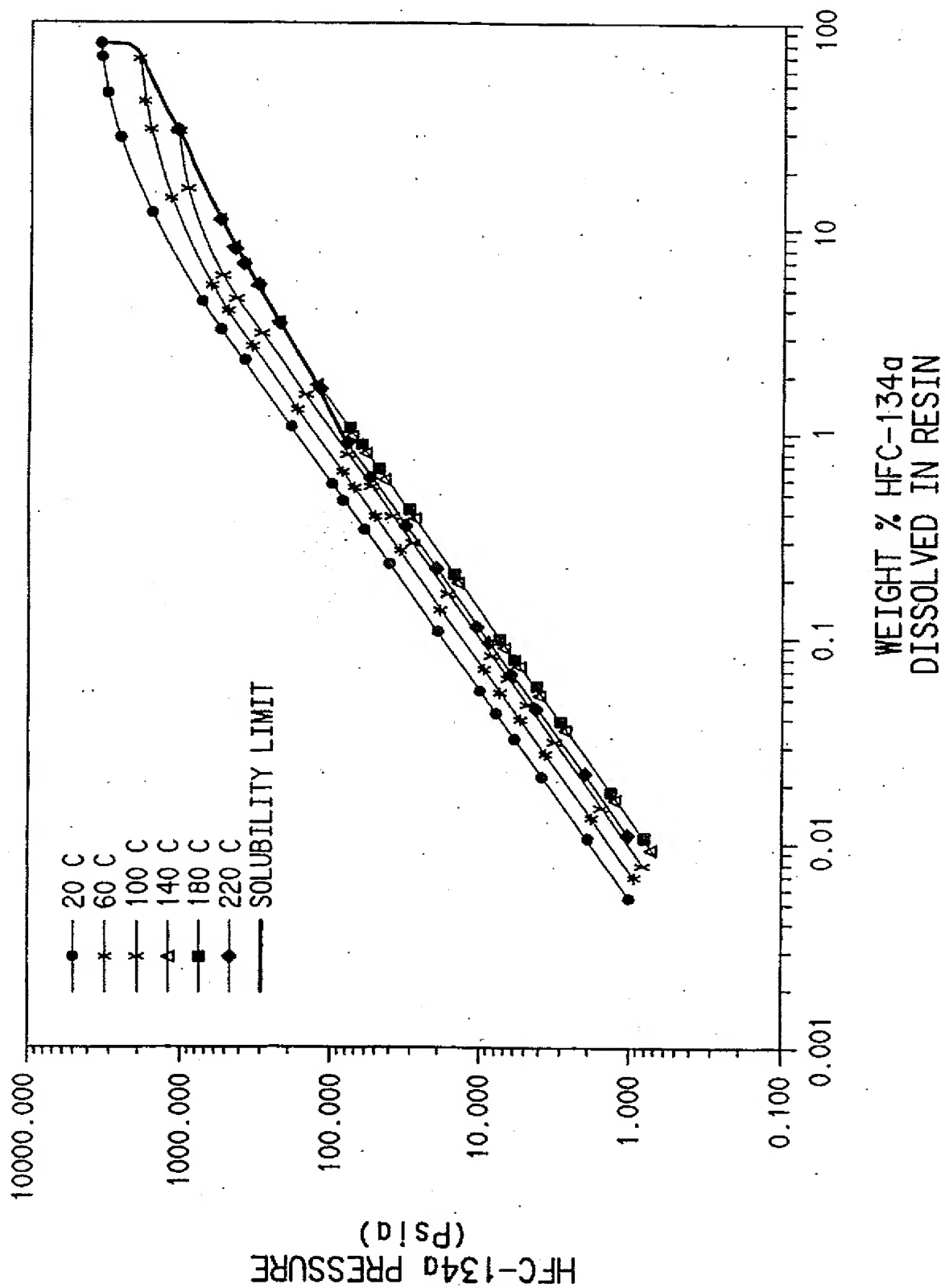
FIG. 3



SUBSTITUTE SHEET (RULE 26)

4/6

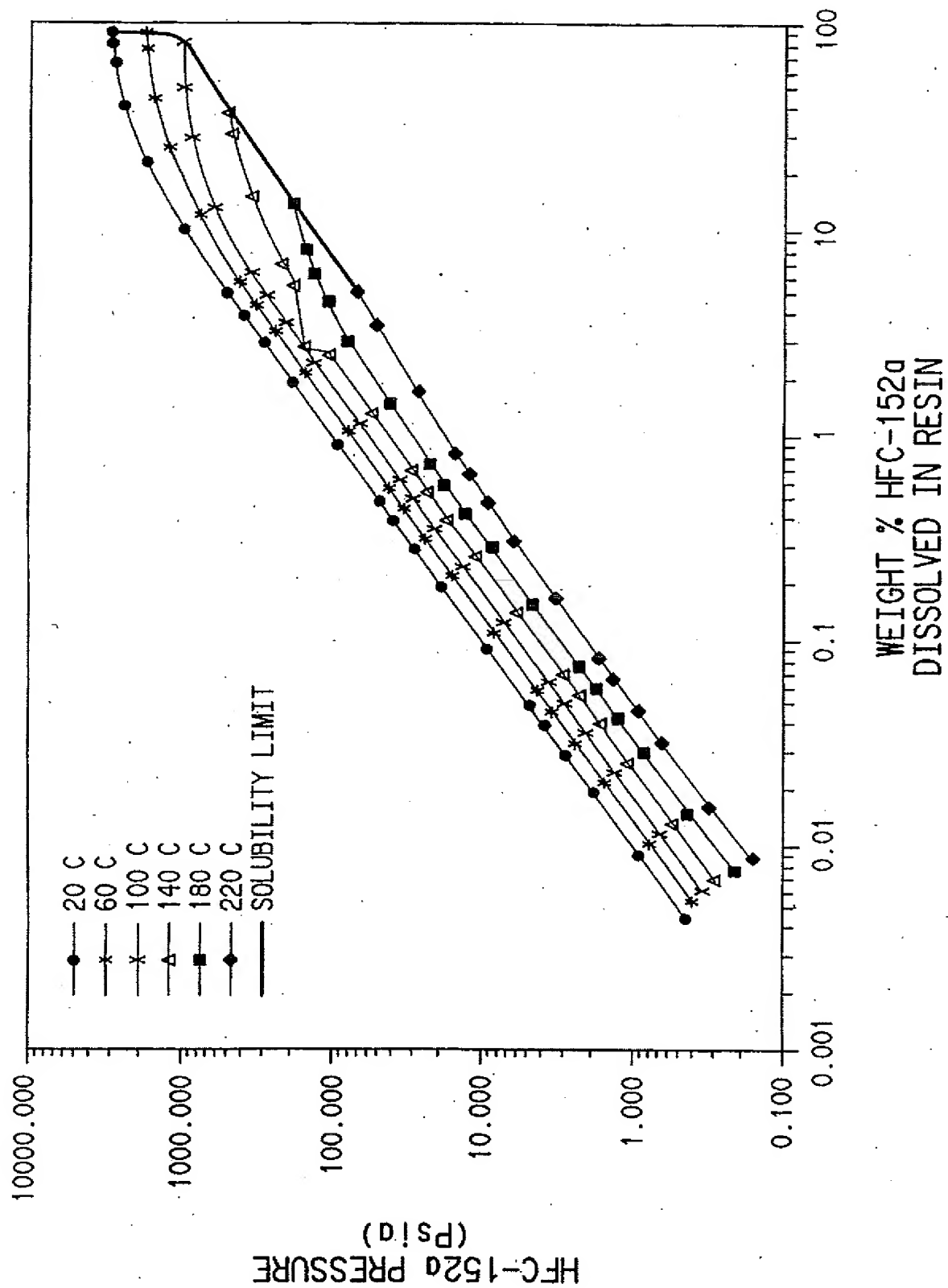
FIG. 4



SUBSTITUTE SHEET (RULE 26)

5/6

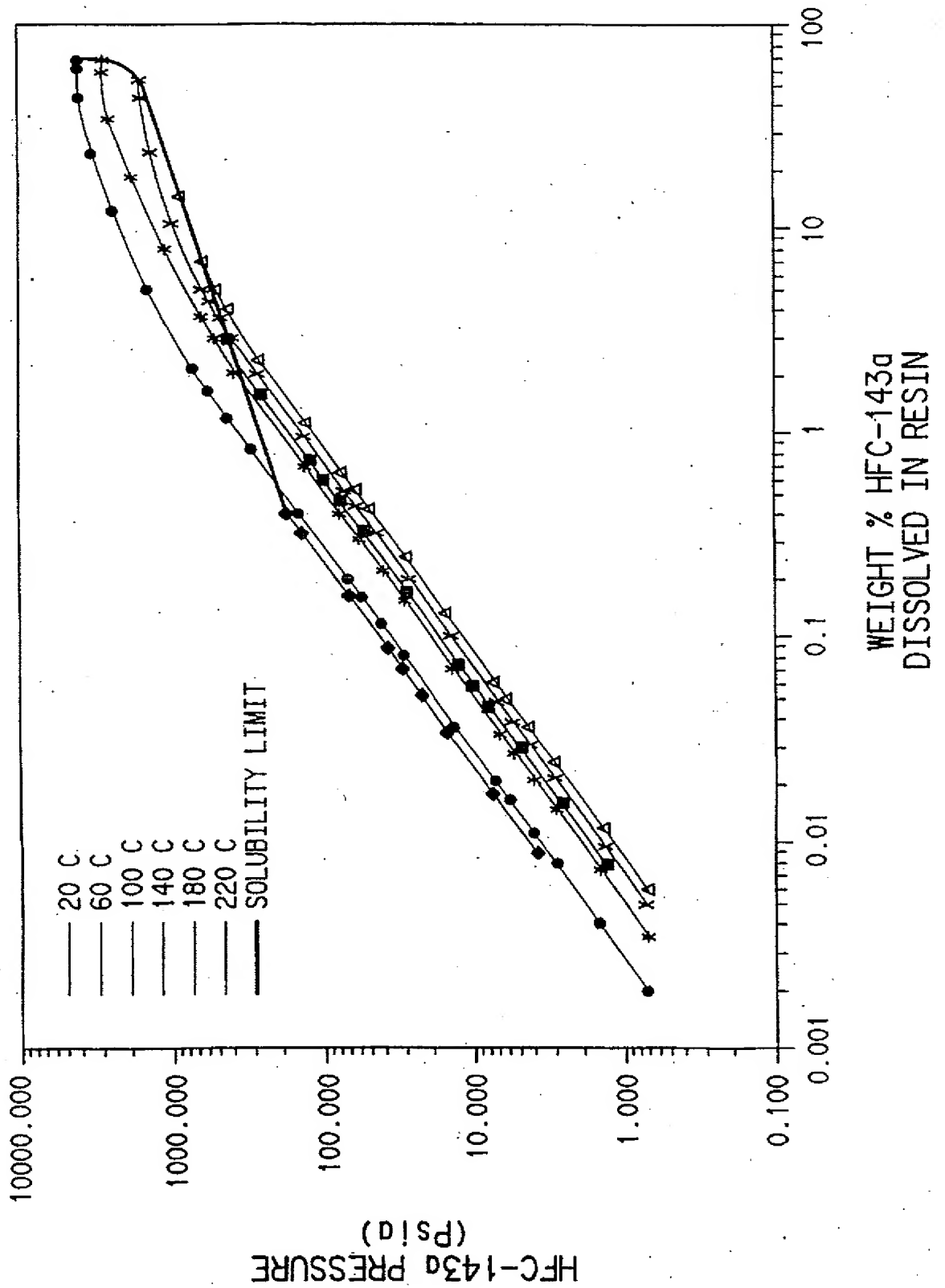
FIG. 5



SUBSTITUTE SHEET (RULE 26)

6/6

FIG. 6



INTERNATIONAL SEARCH REPORT

Inter. Appl. No.

PCT/US 97/13074

A. CLASSIFICATION OF SUBJECT MATTER
IPC 6 C08J9/14

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
IPC 6 C08J

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	EP 0 515 125 A (DU PONT) 25 November 1992 cited in the application	1, 2, 4, 6-8
Y	see claims 1-3, 8	1, 2, 4, 6, 8
Y	WO 92 17558 A (DU PONT) 15 October 1992 cited in the application see page 8; table II see page 11; table III see claims	1, 2, 4, 6, 8
A	WO 93 09199 A (DU PONT) 13 May 1993 see page 18, line 22-25 see claims	1-8

☐ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

* Special categories of cited documents:

- "A" document defining the general state of the art which is not considered to be of particular relevance
- "E" earlier document but published on or after the international filing date
- "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- "O" document referring to an oral disclosure, use, exhibition or other means
- "P" document published prior to the international filing date but later than the priority date claimed

- "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- "&" document member of the same patent family

Date of the actual completion of the international search

18 November 1997

Date of mailing of the international search report

08/12/1997

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,
Fax: (+31-70) 340-3016

Authorized officer

Oudot, R

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/US 97/13074

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP 0515125 A	25-11-92	US 5147896 A	15-09-92
		AU 2141692 A	30-12-92
		BR 9206019 A	27-09-94
		CA 2109626 A	26-11-92
		CN 1066858 A	09-12-92
		EP 0585378 A	09-03-94
		JP 7502287 T	09-03-95
		MX 9202345 A	01-11-92
		WO 9220730 A	26-11-92
		US 5204169 A	20-04-93
WO 9217558 A	15-10-92	US 5182040 A	26-01-93
		AU 652527 B	25-08-94
		AU 1760892 A	02-11-92
		BR 9205818 A	28-06-94
		CA 2107232 A	29-09-92
		EP 0577750 A	12-01-94
		JP 6508825 T	06-10-94
		US 5648017 A	15-07-97
WO 9309199 A	13-05-93	AU 3056092 A	07-06-93
		CN 1071854 A	12-05-93
		EP 0610424 A	17-08-94
		JP 7500608 T	19-01-95
		MX 9206282 A	01-08-93
		US 5589098 A	31-12-96
		US 5290466 A	01-03-94
		ZA 9208410 A	02-05-94

